

Class - X

## ENTRANCE TEST CUM SCHOLARSHIP (SAMPLE PAPER-3)

[Time: 3 Hours]

[Max Marks: 450]

## A.General:

- 1. This booklet is a Question Paper containing 150 questions.
- 2.Blank Papers, Clipboards, Log Tables, slide rules, calculators, cellular phones and electronic gadgets in any form are not allowed to be carried inside the examination hall.
- 3. The answer sheet, a machine-readable optical mark recognition sheet (OMR Sheet), is provided separately.
- 4.DO NOT TAMPER WITH THE OMR OR THE BOOKLET.
- 5. Please fill your roll number correctly in the OMR sheet (answer sheet).
- 6.Both Question Paper and OMR Answer Shee t will be submitted after completion of this examination.

B.Question Paper Format and marking scheme:

- 1. The Question Paper consists of five parts (Part I: MAT, Part II: Physics, Part III: Chemistry, Part IV: Mathematics).
- 2.Each Question carries +3 marks for correct answer and -1 mark for incorrect answer.

## MAT

Directions (1-2) : Read the following information and answer the questions given below. Mohit lives 2 km to the North of Aasif, who lives 2 km to the north of Rohan. Aayush lives 2 km to the South, who lives 4 km to the East of Aasif.

- 1. What is the distance between Mohit and Rohan?
  - (a) 4 km (b) 2 km (c) 6 km (d) 3 km
- 2. What is the distance between Aayush and Rohan?

(a) 2 km (b) 8 km (c) 4 km (d) 18 km

Directions (3-4) : Read the following information carefully and then answer the question based on that. There are five friends named A, B, C, D and E. Everyone likes to play game. The name of games are tennis, hockey, football, baseball and cricket. All are standing in a queue and facing North but not necessarily in the same order.

- D doesn't like to play hockey and football but standing in the middle of queue.
- E likes to play tennis and standing in the right end.
- B is the neighbour of D and A and likes to play football.
- A likes baseball and standing in the left end of queue.
- 3. Who likes to play cricket?
  - (a) A (b) B (c) C (d) D
- 4. Who is second to the right of B?

Directions (5-6) : Insert the missing numerical value in the following questions.







(b) 512

(c) 16

(d) 24

Direction (7) : Complete the given series by choosing correct alternative.

7. 2, 6, 12, 20, 30, 42, (?), 72

(a) 66 (b) 56 (c) 72 (d) 62

Direction (8) : Select the answer from given alternatives, which would be most suitable when paper is unfolded.

8. Problem Figures.



Direction (9) : Find the correct answer which completes the series.

9. Problem Figures.

11.



10. Find out the missing number in the following figure.



(a) 81 (b) 25 (c) 49 (d) None of these Find out how many 2's in the given series? 242526272829303121202223686262729282102

(a) 17 (b) 16 (c) 18 (d) 15

Direction (12) : Read the information and answer the question given below. If A denotes addition, B denotes division, C denotes minus and D denotes multiplication.

- 12. 54 C 10 D 16 B 4 A 8
  - (a) 22 (b) 8 (c) 9 (d) 12

Directions (13) : You are to choose from the four Venn-diagrams that best illustrates the relationship among three given classes or groups in the following question.



- 13. Tennis, Cricket ,Games
- 14. If  $\frac{11y}{10} \frac{9y}{10} = 1$ , then find the value of y.

(a) 18 (b) 16 (c) 4 (d) 5

15. Market price of an article is Rs. 720 and actual price is Rs. 550.80 after two successive discounts. First is 10%, what is the second discount?

(a) 16% (b) 20% (c) 15% (d) 18%

16. Mita is taller than Seema but not as tall as Divya. Seema is taller than Saroj. Divya is not as tall as Reema. Among them who is the tallest?

(a) Reema (b) Saroj (c) Divya (d) Mita

Direction (17) : Find the correct answer, when a sheet having a certain design is folded along the dotted line.

17. Problem Figure



Direction (18) : Find out the figure which is exactly similar with the problem figure.

18. Problem Figure



(a)

21.

22.

23.



Directions (19-20) : Each of these questions is based on following information.

M % N means M is the son of N.

M @ N means M is the sister of N.

M \$ N means M is the father of N.

- 19. Which of the following shows the relation that C is the granddaughter of E?
  - (a) C % B \$ F \$ E (b) B \$ F \$ E % C
  - (c) C @ B % F % E (d) D % B \$ F \$ C

20. Which of the following shows the relation that S is the father of Q?

(a) S @ P \$ Q (b) Q @ P % S (c) Q \$ S @ P

(d) None of these

Directions (21-23) : A code language has been used to write the words in capital letters in English in column I as number letters in column II. Number in column II do not appear in the same order as letters in column I. Decode the language and choose the correct code for the word given in each question from the given alternatives-

		I		11	
		I		11	
		ANT		307	
		TEN		237	
		SUN		345	
		SET		752	
•	Code for 'A' is-				
	(a) 3	(b) 0	(c) 7	(d) 5	
•	Code for word 'NUT' is-				
	(a) 432	(b) 347	(c) 503	(d) 724	
•	Code for letter 'U' is-				
	(a) 3	(b) 0	(c) 5	(d) 4	

Directions (24-27) : Find the relationship between C and D as given between A and B in the problem figure.

24. Problem Figures



28.	Deepu is facing West. He moves 90° in anti-clockwise direction and 135° in clockwise direction. In which direction Deepu is facing now?					
	(a) North-East	(b) South-East	(c) South	(d) North-West		
29.	Pointing to a man in a photograph, a woman said, "He is the only son of my mother-in-law". How is woman related to that man?					
	(a) Sister	(b) Mother-in-law	(c) Wife	(d) None of these		
30.	Today is Monday, what will be the day after 308 days?					
	(a) Sunday	(b) Monday	(c) Tuesday	(d) Thursday		
		PH	YSICS			
31.	•	are first connected in serie the minimum resistance?		in parallel. What is the ratio of		
	(a) n	(b) 1/n <sup>2</sup>	(c) n <sup>2</sup>	(d) 1/n		
32. Two resistances $R_1$ and $R_2$ consume power at the rate of 25 W and 100 W resp connected in series and parallel across the same 120V line. Then the ratio of powe $R_1$ to that consumed by $R_2$ when connected across a 120 V line separately will be				he ratio of power consumed by		
	(a) 1 : 1	(b) 1 : 2	(c) 2 : 1	(d) 1 : 4		
33.	The masses of the three wires of copper are in the ratio of 1 : 3 : 5 and their length are in the ratio of 5 : 3 : 1. Find the ratio of their resistance.					
	(a) 1:3:5	(b) 5 : 3 : 1	(c) 1 : 25 : 125	(d) 125 :15 : 1		
34.	Two resistances are joined in parallel whose resultant is 6/5 ohm. One of the resistance wires is broken and the effective resistance becomes 2 ohm. Then the resistance in ohm of the wire that got broken was:					
	(a) 3/5	(b) 2	(c) 6/5	(d) 3		
35.	In a neon gas discharge tube $2.9 \times 10^{18}$ ions move to the right through a cross-section of the tube each second, while $1.2 \times 10^{18}$ electrons move to the left in this time. The electronic charge is $1.6 \times 10^{-19}$ coulomb. Then the net electric current in the tube is:					
	(a) 1 amp to the	e right	(b) 0.66 amp to tl	ne right		
	(c) 0.66 amp to	the left	(d) Zero			
36.	Two electric bulbs, one of 200 volt – 40 watt and the other of 200 volt–100 watt are connected in a house wiring circuit, then :					
	(a) They have equal currents through them					
	(b) The resistance of the filaments in both the bulbs is same					
	(c) The resistance of the filament in 40 watt bulb is more than the resistance in 100 watt bulb					
	(d) The resistance of the filament in 100 watt bulb is more than the resistance in 40 watt bulb					
37.	Resistance of an ideal voltmeter is :					
	(a) Low	(b) Very low	(c) Infinite	(d) Can't say		
		., ,	7			

38. Figure below shows a balanced wheatstone's network. Now, it is disturbed by changing P to 15  $\Omega$ . Which of the following steps will not bring the bridge to balance again?



42. Six equal resistances, each of 4 ohm, are connected to form the figure shown. The resistance between any two corners is:



(a) 4 ohm

(b) 2 ohm

(c) 1 ohm

(d) 4/6 ohm

43. In the circuit shown below  $R_1 = 10\Omega$ ,  $R_2 = 20\Omega$ ,  $R_3 = 30 \Omega$  and the potentials of points A, B and C are 10V, 6V and 5V respectively. The current through resistance  $R_1$  is:



44. Fig. represents a part of closed circuit. The potential difference between A and B i.e.  $V_A - V_B$  is:



- (a) 24 V(b) 0 V(c) 6 V(d) 18 V45. Which of these is not a safety measure?
  - (a) Fuse (b) MCB (c) Heater (d) Earthing
- 46. Two infinitely long wires carry currents in opposite directions. The magnetic field at a point P lying midway between them is:
  - (a) Twice the field due to each wire alone (b) Half of the field due to each wire alone
  - (c) Square of the field due to each wire alone (d) Zero
- 47. A β-particle moving with a speed of 10<sup>6</sup> ms<sup>-1</sup> enters into the region of a uniform magnetic field of 0.2 T as shown in Fig. The force experienced by the β-particle is:
  - (a)  $2.77 \times 10^{-14}$  N (b)  $1.6 \times 10^{-14}$  N (c)  $5.54 \times 10^{-14}$  N (d)  $2.77 \times 10^{-13}$  N



- 48. A proton enters in a magnetic field of strength B (Tesla) with speed v, parallel to the direction of magnetic lines of force. The force on the proton is:
  - (a) evB (b) Zero (c)  $\infty$  (d) evB/2

49. Three infinite straight wires A, B and C carry currents as shown in Fig. The resultant force on wire B is directed :



- (a) Towards A
- (c) Zero

(b) Towards C

(d) Perpendicular to the plane of the page

- 50. A strong magnetic field is applied on a stationary electron, then:
  - (a) The electron moves in the direction of the field.
  - (b) The electron moves in an opposite direction.
  - (c) The electron remains stationary.
  - (d) The electron starts spinning.
- 51. Electron and proton of equal momentum enter a uniform magnetic field normal to the lines of force. If the radii of curvature of circular paths be r<sub>e</sub> and r<sub>p</sub> respectively, then:

(a) 
$$\frac{r_{e}}{r_{p}} = \frac{1}{1}$$
 (b)  $\frac{r_{e}}{r_{p}} = \frac{m_{p}}{m_{e}}$  (c)  $\frac{r_{e}}{r_{p}} = \sqrt{\left(\frac{m_{p}}{m_{e}}\right)}$  (d)  $\frac{r_{e}}{r_{p}} = \sqrt{\left(\frac{m_{e}}{m_{p}}\right)}$ 

52. A charge (q) passing through a uniform electric field (E) and uniform magnetic field  $(\vec{B})$  remains undeflected. Which of the following variations would still let the charge to remain undeflected? (a) Signs of q and  $\vec{B}$  are changed (b) Signs of q and  $\vec{E}$  are changed

(c) Signs of  $\vec{B}$  and  $\vec{E}$  are changed (d)

(d) None of these

53. The magnetic field lines due to a bar magnet are correctly shown in:





- 54. A person standing in front of a mirror finds his image thinner but with normal height. This implies that the mirror is:
  - (a) Convex and spherical

- (b) Concave and spherical
- (c) Convex and cylindrical with axis vertical (d) Convex and cylindrical with axis horizontal

- 55. A person standing in the centre of a room, looks into a plane mirror fixed on the wall. Then the minimum length of the plane mirror for him to see the full length image of the wall behind him is equal to:
  - (a) Height of the wall (b)  $\frac{2}{3}$ rd of the height of the wall

(c) 
$$\frac{1}{3}$$
 rd of the height of the wall

(d) Half of the height of the wall

56. With a fixed incident ray, if a plane mirror is rotated through an angle  $\theta$  about an axis lying in the plane of incidence, then the reflected ray turns through an angle:

(a) 
$$\theta$$
 (b)  $2\theta$  (c)  $\theta/2$  (d)  $3\theta$ 

57. Two plane mirrors A and B are aligned parallel to each other, as shown in the figure. A light ray is incident at an angle of 30° at a point just inside on end of A. The plane of incidence coincides with the plane of the figure. The maximum number of times the ray undergoes reflections (including the first one) before it emerges out is :



65.	5. A convex lens of focal length 0.1 m is illuminated with a parallel beam of white light. Then, the image obtained at a distance of 0.1 m from the lens is:					
	(a) White	(b) Red coloured	(c) Violet	(d) Yellow		
66.	Which mirror can produce virtual and magnified image?					
	(a) Concave n	nirror (b) Convex mirror	(c) Plane mirror	(d) None of these		
67.	A book looks red when seen through a piece of red glass, then the cover must be of :					
	(a) Red	(b) White	(c) Green	(d) Red or White		
68.	Magnetic lines of force:					
	(a) cannot inte	ersect at all	(b) intersect within the magnet			
	(c) intersect of	nly at south and north poles	(d) intersect at neut	ral points only		
69.	Which of the following is most suitable for the core of the electromagnets?					
	(a) Air	(b) Soft iron	(c) Steel	(d) Cu-Ni alloy		
70. Two thin long parallel wires separated by a distance b are carrying a current i ampere eac magnitude of the force per unit length exerted by one wire on the other is:						
	(a) $\mu_0(i^2 / b^2)$	(b) $\mu_0 i^2 / 2\pi b$	(c) $\mu_0 i / 2\pi b$	(d) μ <sub>0</sub> i / 4πb		
	CHEMISTRY					
74						
/1.	Which substanc	e is used as a leaching agent	in concentration of b	pauxite?		
71.						
71. 72.	(a) H <sub>2</sub> O	e is used as a leaching agent (b) KOH llowing salts is not an alum?	(c) NaOH	oauxite? (d) CaO		
	(a) H <sub>2</sub> O Which of the fo	(b) KOH	(c) NaOH	(d) CaO		
	(a) $H_2O$ Which of the fo (a) $K_2SO_4.AI_2$	(b) KOH llowing salts is not an alum?	(c) NaOH	(d) CaO		
72.	(a) $H_2O$ Which of the fo (a) $K_2SO_4.AI_2$	(b) KOH llowing salts is not an alum? $SO_4)_3 \cdot 24H_2O$ $AI_2(SO_4)_3 \cdot 24H_2O$	(c) NaOH (b) $Na_2SO_4 AI_2(SO_4)$	(d) CaO		
72.	(a) $H_2O$ Which of the fo (a) $K_2SO_4.AI_2($ (c) $(NH_4)_2SO_4$	(b) KOH llowing salts is not an alum? $SO_4)_3.24H_2O$ $AI_2(SO_4)_3.24H_2O$ ase of $HPO_4^{2-}$ is:	(c) NaOH (b) $Na_2SO_4 AI_2(SO_4)$	(d) CaO ) <sub>3</sub> .24H <sub>2</sub> O		
72.	(a) $H_2O$ Which of the fo (a) $K_2SO_4.AI_2O_4$ (c) $(NH_4)_2SO_4$ The conjugate b (a) $PO_4^{3-}$	(b) KOH llowing salts is not an alum? $SO_4)_3.24H_2O$ $AI_2(SO_4)_3.24H_2O$ ase of $HPO_4^{2-}$ is:	(c) NaOH (b) $Na_2SO_4 AI_2(SO_4)$ (d) None of these (c) $H_3PO_4$	(d) CaO ) <sub>3</sub> .24H <sub>2</sub> O (d) H <sub>2</sub> PO <sub>2</sub>		
72. 73.	(a) $H_2O$ Which of the fo (a) $K_2SO_4.AI_2O_4$ (c) $(NH_4)_2SO_4$ The conjugate b (a) $PO_4^{3-}$	(b) KOH llowing salts is not an alum? $SO_4)_3.24H_2O$ $AI_2(SO_4)_3.24H_2O$ ase of $HPO_4^{2-}$ is: (b) $H_2PO_3$	(c) NaOH (b) $Na_2SO_4 AI_2(SO_4)$ (d) None of these (c) $H_3PO_4$	(d) CaO ) <sub>3</sub> .24H <sub>2</sub> O (d) H <sub>2</sub> PO <sub>2</sub>		
72. 73.	(a) $H_2O$ Which of the fo (a) $K_2SO_4.AI_2($ (c) $(NH_4)_2SO_4$ The conjugate b (a) $PO_4^{3-}$ The pair of meta (a) Mg, Cu	(b) KOH llowing salts is not an alum? $SO_4)_3.24H_2O$ $AI_2(SO_4)_3.24H_2O$ ase of $HPO_4^{2-}$ is: (b) $H_2PO_3$ als which will produce hydro	<ul> <li>(c) NaOH</li> <li>(b) Na<sub>2</sub>SO<sub>4</sub>.Al<sub>2</sub>(SO<sub>4</sub>)</li> <li>(d) None of these</li> <li>(c) H<sub>3</sub>PO<sub>4</sub></li> <li>ogen gas on reaction v</li> </ul>	(d) CaO ) <sub>3</sub> .24H <sub>2</sub> O (d) H <sub>2</sub> PO <sub>2</sub> with dilute acid is :		
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<ol> <li>72.</li> <li>73.</li> <li>74.</li> <li>75.</li> <li>76.</li> </ol>	(a) $H_2O$ Which of the form (a) $K_2SO_4 \cdot AI_2(a)$ (c) $(NH_4)_2SO_4$ The conjugate be (a) $PO_4^{3-}$ The pair of meta (a) Mg, Cu What is the nature (a) Acidic Which of the form (a) CO Which of the form (a) Nitric acid	(b) KOH llowing salts is not an alum? $SO_4)_3.24H_2O$ $AI_2(SO_4)_3.24H_2O$ ase of $HPO_4^{2-}$ is: (b) $H_2PO_3$ als which will produce hydro (b) Mg, Ag ure of aqueous ammonia? (b) Basic llowing oxides will turn moi (b) NO llowing substances is used in	(c) NaOH (b) $Na_2SO_4.AI_2(SO_4)$ (d) None of these (c) $H_3PO_4$ ogen gas on reaction v (c) Mg, Zn (c) Neutral st blue litmus red? (c) $N_2O$	(d) CaO ) <sub>3</sub> .24H <sub>2</sub> O (d) H <sub>2</sub> PO <sub>2</sub> with dilute acid is : (d) Cu, Zn (d) Amphoteric (d) P <sub>2</sub> O <sub>5</sub> fuel? kide		

When 0.4 g of NaOH is dissolved in one litre of solution, the pH of solution is: (a) 12 (b) 2 (c) 6 (d) 10						
(a) $CaCO_3 \longrightarrow CaO + CO_2$		$aO + CO_2$	(D) (	$aO + 2HCI \longrightarrow$	$CaCl_2 + H_2O$	
(c)	$Fe + CuSO_4$ —	$\rightarrow$ FeSO <sub>4</sub> + Cu	(d) N	$AaOH + HCl \longrightarrow$	$NaCl + H_2O$	
80. In reaction $SO_2 + 2H_2S \longrightarrow 2H_2O + 3S$ , the reducing agent is :						
(a)	SO <sub>2</sub>	(b) H <sub>2</sub> S	(c) H	l <sub>2</sub> O	(d) S	
In t	In the reaction M + $O_2 \rightarrow MO_2$ (super oxide) the metal is :					
(a)	Li	(b) Na	(c) K		(d) All of these	
The most abundant metal in the earth's crust is :						
(a)	AI	(b) Fe	(c) O		(d) Cu	
How many atoms are contained in a mole of Ca(OH) <sub>2</sub> ?						
					ms/mol	
(c) $6 \times 6.02 \times 10^{23}$ atoms/mol (d) None of these						
The raw materials required for the manufacture of Na <sub>2</sub> CO <sub>3</sub> by Solvay process are:						
				(b) $NH_4CI$ , NaCI, Ca(OH) <sub>2</sub>		
					<sub>3</sub> , H <sub>2</sub> O	
85. Among HCI, HBr, HF, and HI the weakest acid in water is:						
(a)	HCI	(b) HF			(d) HI	
<ul><li>86. Which of the following statements about graphite is incorrect?</li><li>(a) Graphite is a good conductor of electricity.</li></ul>						
(b)						
(c) Graphite is the hardest substance.						
(d)	(d) Graphite is lustrous.					
87. Which of the following statements is incorrect?						
(a) Metals like Cu, Ag, Au cannot displace 'H' from acids.						
(b) In reactivity series metals are arranged in order of increasing reactivity.						
(c)	Silver cannot di	ilver cannot displace 'Cu' from $Cu(NO_3)_2$ .				
(d)	(d) Zinc displaces 'Cu' from $CuSO_4$ .					
In tl	he balanced chem	nical equation :				
$aFe_2O_3 + bH_2 \rightarrow cFe + dH_2Oa$ , b, c and d respectively are :						
(a)	1123	(b) 1111	(c) 13	323	(d) 1223	
	<ul> <li>(a)</li> <li>(b)</li> <li>(c)</li> <li>(c)</li> <li>(c)</li> <li>(c)</li> <li>(c)</li> <li>(c)</li> <li>(c)</li> <li>(d)</li> <li>(c)</li> <li>(d)</li> <li>(d)</li> <li>(d)</li> <li>(c)</li> <li>(d)</li> <li>(d)</li></ul>	(a) 12 Which of the followi (a) $CaCO_3 \longrightarrow C$ (c) $Fe + CuSO_4 \longrightarrow C$ In reaction $SO_2 + 2Fe$ (a) $SO_2$ In the reaction $M + C$ (a) Li The most abundant of (a) Al How many atoms and (a) $3 \times 6.02 \times 10^{23}$ (c) $6 \times 6.02 \times 10^{23}$ The raw materials reaction (a) $CaCl_2$ , $(NH_4)_2CC$ (c) $NaCl$ , $(NH_4)_2CC$ (c) $NaCl$ , $(NH_4)_2CC$ (c) $NaCl$ , $(NH_4)_2CC$ (c) $NaCl$ , $(NH_4)_2CC$ (c) $SaCl_2$ , $(She Charter a)$ (a) $CaCl_3$ , $(She Charter a)$ (b) $Graphite$ is a go (c) $Graphite$ is $She Charter a)$ (c) $Silver$ cannot $di$ (d) $Zinc$ $displaces of a)$ In the balanced cherry $aFe_2O_3 + bH_2 \rightarrow cFe + a)$	(a) 12 (b) 2 Which of the following is a displacement read (a) $CaCO_3 \longrightarrow CaO + CO_2$ (c) $Fe + CuSO_4 \longrightarrow FeSO_4 + Cu$ In reaction $SO_2 + 2H_2S \longrightarrow 2H_2O + 3S$ , the (a) $SO_2$ (b) $H_2S$ In the reaction $M + O_2 \rightarrow MO_2$ (super oxide (a) Li (b) Na The most abundant metal in the earth's cruss (a) AI (b) Fe How many atoms are contained in a mole of (a) $3 \times 6.02 \times 10^{23}$ atoms/mol (c) $6 \times 6.02 \times 10^{23}$ atoms/mol The raw materials required for the manufact (a) $CaCI_2$ , $(NH_4)_2CO_3$ , $NH_3$ Among HCI, HBr, HF, and HI the weakest are (a) HCI (b) HF Which of the following statements about graving (c) Graphite is a good conductor of electric (b) Graphite is the hardest substance. (c) Graphite is lustrous. Which of the following statements is incorrect (d) Graphite is lustrous. Which of the following statements is incorrect (a) Metals like Cu, Ag, Au cannot displace (b) In reactivity series metals are arranged i (c) Silver cannot displace 'Cu' from Cu(NC (d) Zinc displaces 'Cu' from CuSO <sub>4</sub> . In the balanced chemical equation : $aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bH_2 \rightarrow cFe + dH_2O a, b, c and d resplayed aFe_2O_3 + bFe_2O_3 + bFe_2O_3 + bFe_2O_3 + bFe_2O_3 + bFe_2O_3 + bFe_2O_3 + bFe$	(a) 12 (b) 2 (c) 6 Which of the following is a displacement reaction? (a) $CaCO_3 \longrightarrow CaO + CO_2$ (b) C (c) $Fe + CuSO_4 \longrightarrow FeSO_4 + Cu$ (d) M In reaction $SO_2 + 2H_2S \longrightarrow 2H_2O + 3S$ , the reduced (a) $SO_2$ (b) $H_2S$ (c) H In the reaction $M + O_2 \rightarrow MO_2$ (super oxide) the m (a) Li (b) Na (c) K The most abundant metal in the earth's crust is : (a) Al (b) Fe (c) CO How many atoms are contained in a mole of Ca(O (a) $3 \times 6.02 \times 10^{23}$ atoms/mol (b) 5 (c) $6 \times 6.02 \times 10^{23}$ atoms/mol (d) N The raw materials required for the manufacture of (a) $CaCl_2$ , $(NH_4)_2CO_3$ , $NH_3$ (d) N Among HCI, HBr, HF, and HI the weakest acid in w (a) HCI (b) HF (c) H Which of the following statements about graphite if (a) Graphite is a good conductor of electricity. (b) Graphite is the hardest substance. (c) Graphite is lustrous. Which of the following statements is incorrect? (a) Metals like Cu, Ag, Au cannot displace 'H' from (b) In reactivity series metals are arranged in order (c) Silver cannot displace 'Cu' from Cu(NO <sub>3</sub> ) <sub>2</sub> . (d) Zinc displaces 'Cu' from CuSO <sub>4</sub> . In the balanced chemical equation : $aFe_2O_3 + bH_2 \rightarrow cFe + dH_2Oa$ , b, c and d respectively	(a)12(b)2(c)6Which of the following is a displacement reaction?(a) $CaCO_3 \longrightarrow CaO + CO_2$ (b) $CaO + 2HCI \longrightarrow CaO + CO_2$ (c)(c) $Fe + CuSO_4 \longrightarrow FeSO_4 + Cu$ (d) $NaOH + HCI \longrightarrow CaO + 2H_2S \longrightarrow 2H_2O + 3S$ , the reducing agent is :(a) $SO_2$ (b) $H_2S$ (c) $H_2O$ In the reaction $SO_2 + 2H_2S \longrightarrow 2H_2O + 3S$ , the reducing agent is :(a)(a) $Li$ (a) $SO_2$ (b) $H_2S$ (c) $H_2O$ In the reaction $M + O_2 \rightarrow MO_2$ (super oxide) the metal is :(a)Li(b)Na(c)(a)Li(b)Na(c)KThe most abundant metal in the earth's crust is :(a)AI(b)Fe(c)(c)How many atoms are contained in a mole of $Ca(OH)_2$ ?(a)3 × 6.02 × 10^{23} atoms/mol(b)5 × 6.02 × 10^{23} atom(c)6 × 6.02 × 10^{23} atoms/mol(d)None of theseThe raw materials required for the manufacture of $Na_2CO_3$ by Solva(a)CaCl <sub>2</sub> , $(NH_4)_2CO_3$ , $NH_3$ (d)NaCl, $NH_2$ , $CaCO_3$ (a)McI, $(NH_4)_2CO_3$ , $NH_3$ (d)NaCl, $NH_3$ , $CaCO_3$ (b)HF(c)HBrWhich of the following statements about graphite is incorrect?(a)Graphite has a high melting point.(c)Graphite is lustrous.Which of the following statements is incorrect?(a)Metals like Cu, Ag, Au cannot displace 'H' from acids.(b)	

89.	In th	ne reaction :				
	$2\text{FeCl}_3 + \text{SnCl}_2 \rightarrow 2\text{FeCl}_2 + \text{SnCl}_4$					
	(a) Fe <sup>+3</sup> is reduced to Fe <sup>+2</sup>		(b) Sn <sup>+2</sup> is reduced to Sn <sup>+4</sup>			
	(c)	Sn <sup>+2</sup> is oxidized	to Sn	(d) Fe <sup>+3</sup> gains two electrons		
90.	Heat of neutralisation is less than 13.7 kcal/mol for the reaction :					
	(a)	(a) HCI + NaOH $\rightarrow$ NaCI + H <sub>2</sub> O		(b) H <sub>2</sub> SO <sub>4</sub> + 2NaOH –		
	(c) $HNO_3 + NaOH \rightarrow NaNO_3 + H_2O$		(d) CH <sub>3</sub> COOH + NaO	(d) $CH_3COOH + NaOH \rightarrow CH_3COONa + H_2O$		
91.	pH of tomato juice is 4, that means it is :					
	(a)	basic	(b) acidic	(c) neutral	(d) none of these	
92.	Which of the following salts can produce only two types of radicals?					
	(a)	NaKCO <sub>3</sub>	(b) CaOCI <sub>2</sub>	(c) NH <sub>4</sub> HSO <sub>4</sub>	(d) Na(NH <sub>4</sub> )HPO <sub>4</sub>	
93.	Exp	anded octet occur	s in :			
	(a)	NH <sub>3</sub>	(b) PF <sub>5</sub>	(c) H <sub>2</sub> O	(d) O <sub>2</sub>	
94.		•	•		e which slowly disappears	
			ve can feel a pungent od			
	(a)	carbon dioxide	(b) sulphur dioxide	(c) sulphur gas	(d) sulphuric acid	
95.	What process would you employed to extract metal from calamine ore?					
	(a) Concentration, Roasting, Reduction, Refining					
	(b) Concentration, Calcination, Reduction, Refining					
	(c) Concentration, Calcination, Heating, Refining					
	(d) None of these					
96.	The process employed for the purification of copper with cuprous oxide as the impurity is :					
	(a) poling		(b) liquation			
	(C)	electrolytic proce	ess	(d) oxidation		
97.	Calculate the Ka value of a 0.2 M aqueous solution of propanoic acid, $CH_3CH_2CO_2H$ , with a pH of					
	4.88					
	(a)		(b) Ka = $7.69 \times 10^{-10}$		(d) Ka = $5.69 \times 10^{-10}$	
98.	Identify the compound that is the final product of Solvay process.					
	(a) Ammonium carbonate		(b) Ammonium bicarbonate			
00	(C)	Sodium carbona		(d) Sodium bicarbona		
99.	Which of the following gases with a smell of rotten egg is evolved when sulphuric acid is treated with copper sulphide?					
	(a)	Sulphur vapour		(b) Sulphur dioxide		

(c) Hydrogen sulphide (d) Sulphur trioxide

100. Select the steps required for the extraction of aluminium and sequence them. (1) froth floatation (2) chemical separation (3) electrolytic reduction (4) smelting (5) distillation under low pressure (6) electrolytic refining (7) Pulverisation (8) conversion of concentrated ore into oxide (9) crushing and grinding 95723861 (a) (b) 3 2 1 4 5 6 7 8 (c) 972836 (d) 972386 101. Plaster of Paris is obtained : by adding water to calcium sulphate. (a) by adding sulphuric acid to calcium hydroxide. (b) by heating gypsum to a very high temperature. (c) (d) by heating gypsum to 373 K. 102. The correct order of acidic strength is : (a)  $CaO < CuO < H_2O < CO_2$ (b)  $H_2O < CuO < CaO < CO_2$ (d)  $H_2O < CO_2 < CaO < CuO$ (c)  $CaO < H_2O < CuO < CO_2$ 103. Which of the following turns red litmus blue? (a) Water (b) Vinegar (c) Lime water (d) Brine 104. Which of the following metals react with NaOH as well as HCI? (a) Na (b) Ca (c) AI (d) Mg 105. What is pH of 0.01 M monobasic acid solution? (a) 2 (b) 1 (c) 3 (d) 4 106. Which of the following gases is least reactive? (a) N<sub>2</sub> (b) H<sub>2</sub> (c) Cl<sub>2</sub> (d) O<sub>2</sub> 107. Ammonia gas is formed by the combination of nitrogen and hydrogen  $N_2 + 3H_2 \rightarrow 2NH_3$ Which of the following statements is not correct about above equation? (a) Nitrogen and hydrogen are reactants. (b) One molecule of nitrogen combine with 3 molecules of hydrogen and forms two molecules of ammonia. (c) One volume of nitrogen and three volume of hydrogen combine and give 2 volume of ammonia gas. (d) Reactants and products are not gaseous. 108. Which of the following salts on dissolving in water, will give a solution with pH less than 7 at 298 K? (a) KCN (b) CH<sub>2</sub>COONa (c) NaBr (d) NH<sub>4</sub>CI

- 109. Which of the following statements is/are true?
  - (a) The total mass of the substance remains same in a chemical change.
  - (b) A chemical change is permanent and irreversible or reversible.
  - (c) A physical change is temporary and reversible.
  - (d) All of these.

110. Galena is an ore of : (a) Lead (b) Silver (c) Iron (d) Aluminium MATHEMATICS 111. If a, b, and c are the solutions of the equation  $x^3 - 3x^2 - 4x + 5 = 0$ , find the value of  $\frac{1}{ab} + \frac{1}{bc} + \frac{1}{ca}$ (a) -3/4 (b) 3/5 (c) -4/5(d) - 3/5112. The value of  $\cos^2 5^\circ + \cos^2 10^\circ + \cos^2 15^\circ + \dots + \cos^2 90^\circ$  is: (b)  $8\frac{1}{2}$ (d)  $2\frac{1}{2}$ (a) 0 (c) 10 113. In a number of two digits, unit's digit is twice the tens digit. If 36 is added to the number, the digits are reversed. The number is : (a) 36 (b) 63 (c) 48 (d) 84 114. If  $x = p \sec \theta$  and  $y = q \tan \theta$  then : (a)  $x^2 - y^2 = p^2 q^2$ (b)  $x^2q^2 - y^2p^2 = pq$ (c)  $x^2q^2 - y^2p^2 = \frac{1}{p^2q^2}$ (d)  $x^2q^2 - y^2p^2 = p^2q^2$ 115. If  $f(x) = 2x^4 - 13x^2 + ax + b$  is divisible by  $x^2 - 3x + 2$ , then (a, b) =(b) (6, 4) (d) (2, 9) (a) (-9, -2) (c) (9, 2) 116. If  $\alpha$ ,  $\beta$ ,  $\gamma$  are roots of  $x^3 + 4x + 1 = 0$ , then the equation whose roots are  $\alpha^2 / (\beta + \gamma)$ ,  $\beta^2 / (\gamma + \alpha)$  and  $\gamma^2/(\alpha + \beta)$  can be : (a)  $x^3 - 4x - 1 = 0$ (c)  $x^3 + 4x - 1 = 0$ (b)  $x^3 - 4x + 1 = 0$ (d)  $x^3 + 4x + 1 = 0$ 117. The probability of choosing randomly a number c from the set {1, 2, 3, ....., 9} such that quadratic equation  $x^2 + 4x + c = 0$  has real roots is : (a) 1/9 (c) 3/9 (b) 2/9 (d) 4/9 118. If  $x^2 = 3x - 1$ , then the value of  $\frac{x^2 + 1}{x^3}$  is : (a) 17 (b) 18 (c) 19 (d) 20 119.  $\left(1-\frac{1}{n}\right)\left(1-\frac{1}{n+1}\right)\left(1-\frac{1}{n+2}\right)\cdots\left(1-\frac{1}{2n}\right)$  is : (a)  $\frac{1}{2n}(n-1)$  (b)  $\frac{1}{2n}$ (c)  $\frac{1}{n}$ (d)  $\frac{2n}{n-1}$ 120. In the figure, area of circle is 50sq. cm and the area of triangle is 15 sq. cm, then  $\sin\theta + \sin\alpha + \sin\beta = \dots$ 



(d) None of these

(C) 6π

(b)  $\frac{3\pi}{5}$ 

(a)  $\frac{9\pi}{10}$ 

121. When  $2^{256}$  is divided by 17, the remainder would be :

(a) 10 (b) 16 (c) 14 (d) None of these

122. OPQ is a quadrant of a circle with centre O and semicircles are drawn on it, as shown in figure, then A : B =



124. Mean of n numbers  $x_1, x_2, \dots, x_n$  is m. If  $x_n$  is replaced by x, then the new mean is :

(a) 
$$m - x_n + x$$
 (b)  $\frac{mn - x_n + x}{n}$  (c)  $\frac{(n-1)m + x}{n}$  (d)  $\frac{m - x_n + x}{n}$ 

125. On her vacation Deepika visits four cities A, B, C and D in a random order. What is the Probability that she visits A before B?

(a) 
$$\frac{3}{4}$$
 (b)  $\frac{1}{2}$  (c)  $\frac{5}{24}$  (d)  $\frac{1}{4}$ 

126. If sinx + cosx -  $\sqrt{2}$  sinx = 0, then value of tan<sup>2</sup>x + cot<sup>2</sup>x is :

- (a) 8 (b) 6 (c) 1 (d)  $\frac{1}{2}$
- **127.** ABC is an equilateral triangle. With side AC as diameter a semicircle is drawn as shown in the figure. If OB = 1 cm, then what is length of side of an equilateral triangle. (Given : O is the farthest point from B on the semicircle)



128. If the roots of a quadratic equation are  $\frac{p}{q}$ ,  $-\frac{q}{p}$ , then the equation is : (b)  $pqx^2 - (p^2 - q^2)x - pq = 0$ (a)  $qx^2 - (q^2 + p^2)x - pq = 0$ (d)  $p^2x^2 - (p^2 - q^2)x - pq = 0$ (c)  $px^2 - (p^2 + 1)x + p = 0$ 129. The angle of elevation of a cloud from a point 'h' metre above a lake is  $\theta$ . The angle of depression of its reflection in the lake is 45°. The height of the cloud is : (a)  $h\left(\frac{1+\tan\theta}{1-\tan\theta}\right)$  (b)  $h\left(\frac{1-\tan\theta}{1+\tan\theta}\right)$  (c)  $h\left(\frac{\tan\theta}{1+\tan\theta}\right)$  (d) None of these 130. p, q, r, s are four positive integers such that the product p.q.r.s is odd. If  $x = p^n + q^n$  and  $y = q^n + r^n + s^n$ , then  $(-1)^x + (-1)^y =$ \_\_\_\_ (where n is natural number) (b) -2 (a) 0 (c) 2 (d) can't be determined 131. L.C.M. of  $2017^{2017} - 1$ , and  $2017^{2017} + 1$  is ..... (a)  $2017^{4034} + 1$  (b)  $(2017)^{4034} - 1$  (c)  $\frac{1}{2}(2017^{4034} - 1)$  (d)  $\frac{4034^{4034} - 1}{2}$ 132.  $\sqrt{-\sqrt{3} + \sqrt{3 + 8\sqrt{7 + 4\sqrt{3}}}} =$ (b) 3 (c) 2 (d) 0 (a) 1 133. If  $\sin x + \sin^2 x = 1$ , then the value  $\cos^2 x + \cos^4 x$  is : (a) 1 (b) 2 (c) 0 (d) -1 134. If the polynomial  $ax^3 + bx - c$  is exactly divisible by  $x^2 + bx + c$ , then  $\frac{ac}{b} + ab$  can be : (a) -1 (b) 3 (c) 1 (d) 0 135. If  $7^{1/3} + 49^{1/3}$  is a root of cubic equation, then the product of all the roots of that equation is : (b) 49 (c) -49 (a) 7 (d) 56 136. The minimum value of  $2x^2 - 3x + 5$  is : (c)  $\frac{31}{16}$ (d)  $\frac{29}{16}$ (a)  $\frac{31}{8}$ (b)  $\frac{29}{9}$ 137. Consider the following two sets of equations 1. 2x - y = 0 and 6x - 3y = 0II. 3x - 4y = 0 and 12x - 20y = 0, then : (a) both sets I and II has unique solutions (b) set I has unique solution and set II has infinitely many solutions

- (c) set II has unique solution and set I has infinitely many solutions
- (d) none of the sets I and II has unique solution.
- 138. In the figure a right circular cylinder just encloses a sphere of radius r. Find curved surface area of the cylinder.



(a)  $2\pi r^2$  (b)  $6\pi r^2$ 

(d) None of these

139. In the given figure, ABCD is a rectangle, segments AL and AM are drawn as shown. Then the length of (AL + AM) is :



148. The condition that one root of the equation  $ax^2 + bx + c = 0$ ,  $a \neq 0$  may be double of the other is (a)  $b^2 = 9ac$  (b)  $2b^2 = 9ac$  (c)  $2b^2 = ac$  (d)  $b^2 = ac$ 

- 149. If the zeros of the polynomial  $f(x) = x^3 6x^2 + x + 10$  are a b, a, a + b, then the value of b =(a)  $\pm 1$ (b)  $\pm 2$ (c)  $\pm 3$ (d) None of these
- 150. In the given figure, OPQR is a rhombus three of whose vertices are on the circle with centre O. If the area of the rhombus is  $32\sqrt{3}$  cm<sup>2</sup>, then radius of circle is:



(a) 6 cm

(b) 2 cm

(c) 4 cm

(d) 8 cm